

① ②

$$\frac{\Lambda}{n_H^2} \approx 5 \times 10^{-25} \frac{\text{erg cm}^3}{\text{s}}$$

$$\Lambda = \left(5 \times 10^{-25} \frac{\text{erg cm}^3}{\text{s}} \right) \left(0.5 \text{ cm}^{-3} \right)^2$$

$$\Lambda = 1.25 \times 10^{-25} \frac{\text{erg}}{\text{s cm}^3}$$

$$\begin{aligned} \text{VOLUME OF CLOUD} &= \frac{4}{3} \pi (10 \text{ pc})^3 \left(\frac{3.1 \times 10^{18} \text{ cm}}{\text{pc}} \right)^3 \\ &= 1.25 \times 10^{59} \text{ cm}^3 \end{aligned}$$

$$\begin{aligned} \text{COOLING RATE} &= \left(1.25 \times 10^{-25} \frac{\text{erg}}{\text{s cm}^3} \right) \left(1.25 \times 10^{59} \text{ cm}^3 \right) \\ &= 1.6 \times 10^{34} \frac{\text{erg}}{\text{s}} = 1.6 \times 10^{27} \frac{\text{J}}{\text{s}} \end{aligned}$$

$$L_0 = 3.8 \times 10^{26} \text{ W} \rightarrow \text{SIMILAR!}$$

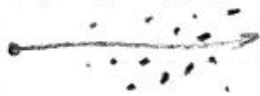
② A GIVEN TEMPERATURE, ELECTRONS WILL MOVE FASTER THAN OTHER MORE MASSIVE PARTICLES

$$\frac{3}{2} kT \approx \frac{1}{2} m v^2$$

AS SUCH, THEY "SWEEP OUT" MORE SPACE AND COLLIDE MORE OFTEN.

THE NUMBER OF COLLISIONS IS THEREFORE VERY SENSITIVE TO THE NUMBER OF ELECTRONS. MORE COLLISIONS = MORE OPPORTUNITIES TO EXCITE ATOMS INTO STATES WHERE THEY CAN RADIATE ENERGY AWAY

③ COLLISIONS MATTER; SEE ABOVE. IMAGINE ONE PARTICLE MOVING THROUGH A CLOUD OF PARTICLES



THE RATE AT WHICH IT COLLIDES IS PROPORTIONAL TO THE DENSITY OF THE CLOUD; THAT'S ONE FACTOR. THE NUMBER OF PARTICLES MOVING TO COLLIDE GIVES YOU THE OTHER FACTOR.

2) a) $d = vt$

$$t = \frac{d}{v} = \frac{120 \text{ mi}}{60 \text{ mi/h}} = 2 \text{ h}$$

b) $\Delta v = at$

$$(0 - 60 \text{ mph}) = (-20 \frac{\text{mph}}{\text{s}}) t$$

$$t = 3 \text{ s}$$

c) $\tau = \frac{\rho}{\lambda}$

$$\frac{\frac{\text{erg}}{\text{cm}^3}}{\frac{\text{erg}}{\text{cm}^3 \cdot \text{s}}} = \text{s} \Rightarrow \text{RIGHT UNITS}$$

3) a) $\rho = (2) \left(\frac{3}{2} kT \right) n_H = \frac{\text{ENERGY PER PARTICLE}}{\text{PARTICLES VOL}} = \frac{\text{ENERGY}}{\text{VOL}}$

EACH H-ATOM
GIVES ONE PROTON
& ONE ELECTRON \rightarrow
2 PARTICLES
(H WILL BE
FULLY IONIZED
@ $T = 10^7 \text{ K}$)

$$\rho = 2 \left(\frac{3}{2} \right) (1.38 \times 10^{-16} \frac{\text{erg}}{\text{K}}) (10^7 \text{ K}) (1 \text{ cm}^{-3})$$

$$\rho = 4.1 \times 10^{-9} \frac{\text{erg}}{\text{cm}^3}$$

b) $\frac{\lambda}{n_H} \sim 2 \times 10^{-23} \frac{\text{erg cm}^3}{\text{s}}$ so $\lambda = 2 \times 10^{-23} \frac{\text{erg}}{\text{cm}^3 \text{ s}}$
FOR $n_H = 1 \text{ cm}^{-3}$

$$\tau = \frac{\rho}{\lambda} = \frac{4.1 \times 10^{-9} \frac{\text{erg}}{\text{cm}^3}}{2 \times 10^{-23} \frac{\text{erg}}{\text{cm}^3 \cdot \text{s}}} = 2 \times 10^{14} \text{ s} = 6 \times 10^6 \text{ yr}$$

3 (c) NUMBERS ARE ALL $\sim 10 \times$ TOO BIG

4 $n_H \sim 0.1 \text{ cm}^{-3}$
 $T \sim 10^7 \text{ K}$

$$g \sim 2 \left(\frac{3}{2} kT \right) n_H$$

$$\sim 2 \left(\frac{3}{2} \right) (1.38 \times 10^{-16} \frac{\text{erg}}{\text{K}}) (10^7 \text{ K}) (0.1 \text{ cm}^{-3})$$

$$\sim 4 \times 10^{-10} \frac{\text{erg}}{\text{cm}^3}$$

$$\Lambda \sim (2 \times 10^{-23} \frac{\text{erg cm}^3}{\text{s}}) (0.1 \text{ cm}^{-3})^2$$

$$\sim 2 \times 10^{-25} \frac{\text{erg}}{\text{cm}^3 \cdot \text{s}}$$

$$\tau \approx \frac{g}{\Lambda} = \frac{4 \times 10^{-10} \frac{\text{erg}}{\text{cm}^3}}{2 \times 10^{-25} \frac{\text{erg}}{\text{cm}^3 \cdot \text{s}}} = 2 \times 10^{15} \text{ s}$$

$$= 7 \times 10^7 \text{ yr}$$

(ABOUT 100 mil yrs)

5 (a) $F_{\text{pix}} = \frac{N_* L_{\text{RG}}}{4\pi d^2}$ # STARS IN ONE PIXEL

(b) $\Delta F_{\text{pix}} = \Delta N_* \frac{L_{\text{RG}}}{4\pi d^2}$

$$= \sqrt{N_*} \frac{L_{\text{RG}}}{4\pi d^2}$$

(c) $\frac{\Delta F_{\text{pix}}}{F_{\text{pix}}} = \frac{\Delta N_*}{N_*} = \frac{\sqrt{N_*}}{N_*} = \frac{1}{\sqrt{N_*}} \Rightarrow N_* = \left(\frac{F_{\text{pix}}}{\Delta F_{\text{pix}}} \right)^2$

$$F_{\text{pix}} = \frac{\left(\frac{F_{\text{pix}}}{\Delta F_{\text{pix}}} \right)^2 L_{\text{RG}}}{4\pi d^2}$$

$$d = \sqrt{\frac{F_{\text{pix}} L_{\text{RG}}}{4\pi (\Delta F_{\text{pix}})^2}}$$

CHECK UNITS

$$\left(\frac{\text{erg}}{\text{cm}^2 \cdot \text{s}} \right) \left(\frac{\text{erg}}{\text{s}} \right)$$

$$\left(\frac{\text{erg}}{\text{cm}^2 \cdot \text{s}} \right) \left(\frac{\text{erg}}{\text{cm}^2 \cdot \text{s}} \right)$$

= 5 ✓